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Chapter 12 - Stoichiometry Note Taking Guide: Episode 801 Stoichiometry -Study of the QUANTITY relationships in a CHEMICAL REACTION -Based on BALANCED equations  $2 \text{Mg} + \text{O}_2 \rightarrow 2 \text{MgO}$  1. Particles 2 atoms Mg 2 atoms O 2 atoms Mg 2 atoms O 2 atoms Mg 1 molecule  $\text{O}_2$  2 molecule MgO 10 atoms Mg 5 molecule  $\text{O}_2$  10 molecule MgO 2 ( $6.02 \times 10^{23}$ ) 1 ( $6.02 \times 10^{23}$ ) atoms Mg molecule  $\text{O}_2$  2 ( $6.02 \times 10^{23}$ ) molecule MgO 2.

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## Access Free Note Taking Guide Episode 801 Stoichiometry Answers

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