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CHAPTER 8 REVIEW
Chemical Equations
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3 SHORT ANSWER

Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg.

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synthesis reaction (8.2)
a reaction in which two
or more substances
combine to form a new

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compound.

decomposition reaction
(8.2) a reaction in
which a single
compound breaks
down to form two or
more simpler
substances.

electrolysis (8.2) the
decomposition of a
substance by an
electric current.

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combination of sulfur

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and another element.

Metal Oxides (8.2)

simple synthesis

reaction - metal and
oxygen -formula MO

(metal-oxygen) Group

2 tend to be MO. Group

1 tend to be M^2O .

(iron can do both)

Metal Sulfides (8.2)

group 1 and 2

elements react with

sulfur to create

sulfides. Group 1

M^2S .

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in certain coordination
complexes.
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Molecular nitrogen (N_2) can act as a ligand in certain ...

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CHAPTER 8 REVIEW

Class sTUò hemical
Equations and

Reactions SECTION 3

SHORT ANSWER

Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg. 2. Consider the metals iron and silver, both listed in Table 3 on page 286 of the

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text. Which one

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Heating 2.0 mol of a
coordination compound
gives 1.0 mol NH_3 ,
2.0 mol H_2O , 1.0 mol

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HCl, and 1.0 mol (NH 4)
3 [Ir 2 Cl 9].
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Heating 2.0 mol of a coordination compound gives 1.0 mol ...

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Propane has the structure $\text{H}_3\text{C} - \text{CH}_2 - \text{CH}_3$. Use

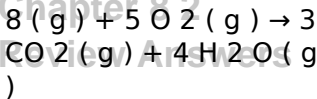
average bond enthalpies from Table

12.5 to estimate the change in enthalpy ΔH°

for the reaction C_3H_8

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Propane has the structure H 3 C — CH 2 — CH 3 . Use ...

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Cobalt(II) forms more tetrahedral complexes than any other ion except zinc(II). Draw the structure(s) of the tetrahedral complex [CoCl 2 (en)] . Could this complex exhibit

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geometric or optical isomerism? If one of the Cl - ligands is replaced by Br - , what kinds of isomerism, if any, are possible in the resulting compound?

Cobalt(II) forms more tetrahedral complexes than any other ...

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Predict the geometry

and number of

unpaired spins in the

following four-

coordinate complexes:

[NiCl₄]²⁻ .

**Predict the
geometry and
number of unpaired
spins in the ...**

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**a. Which is
nonpolar: CF₄ or
CH₂F₂ b. Which
substance ...**

1.7 Chapter

Homework: 1.8

References: Chapter 1

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otherwise noted: 1.

Anonymous. (2012)

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Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is ...

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Chapter 11 - An
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Chemistry: Modern
Atomic Theory ... You
will see in the next
section that the 2 s and
2 p orbitals have

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different potential ...
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