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### **Chapter 12 Review Solutions**

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SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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Modern Chemistry 105 Solutions CHAPTER 12 REVIEW Solutions SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the

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Chapter 12 Review Solutions. 1) 56. 2)  $= 4, = 10, = 16, = 22, = 28$ . 3)  $= 4, = 8, = , = 64, =$ . 4)  $a_n = 2n - 3$ . 5)  $= \cdot 4n$ . 6)  $= 5, = 3, = 1, = -1$ . 7)  $3 + 4 + 5 + \dots + (n + 2) 8$

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CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER

Answer the following questions in the space provided. 1.

Describe the errors made by the following students in making molar solutions.

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Answer the following questions in the space provided. 1. Match  
the type of mixture on the left to its representative particle  
diameter on the right. c solutions (a) larger than 1000 nm a  
suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1  
nm 2. Identify the solvent in each of the following examples:

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Answer the following questions in the space provided. 1. Solid CaCl<sub>2</sub> does not conduct electricity. Explain why it is considered to be an electrolyte. 2. Explain the following statements at the molecular level: a. Generally, a polar liquid and a nonpolar liquid are immiscible. b.

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b. water c. the gels 3. The mixture is a colloid. Modern Chemistry  
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## Chapter 12

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CHAPTER . 12 . REVIEW Solutions . SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

### **CHAPTER REVIEW Solutions**

Modern Chemistry Chapter 12-. Solutions. Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution.

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